625 MSc FORENSIC SCIENCE (FINAL)

- 1. Which one of the following pairs represent units of the same physical quantity?
 - (A) Kelvin and Joule
 - (B) Newton and Calorie
 - (C) Kelvin and Calorie
 - (D) Joule and Calorie
- 2. A force $\vec{F} = \hat{\imath} 2\hat{\jmath} + 3\hat{k}$ acting on a body produces a displacement $\vec{S} = 2\hat{\imath} + 3\hat{\jmath} + 4\hat{k}$. The work done then is
 - (A) 16 units
 - (B) 8 units
 - (C) 32 units
 - (D) Zero
- 3. Planck's constant has the same dimension as
 - (A) Force \times Time
 - (B) Force × Distance
 - (C) Force \times Speed
 - (D) Force × Distance × Time
- 4. The force (*F*) experienced by a charge *q* situated in an electric field of strength (*E*) is given

(A)
$$F = qE$$

(B) $F = q^{2}E$
(C) $F = \frac{E}{q}$
(D) $F = \frac{kE^{2}}{8\pi}$

- 5. The momentum carried by a photon of energy 1 MeV will be
 - (A) 5×10^{-22} kg m/s (B) 0.33×10^{6} kg m/s
 - (b) 0.33×10 kg m/s
 - (C) 7×10^{-24} kg m/s

(D)
$$10^{-22}$$
 kg m/s

6.

- The de-Broglie wavelength of a moving particle of mass *m* and energy *E* is
 - (A) $\sqrt{2mE}$ (B) $\frac{h}{\sqrt{2mE}}$ (C) $\frac{h}{\sqrt{2mET}}$ (D) $\frac{h}{mE}$
- 7. Diffraction of electrons was demonstrated by
 - (A) Davidson and Germer's experiment
 - (B) Michelson-Morley experiment
 - (C) Stern and Gerlach experiment
 - (D) Frank-Hertz experiment
- 8. Two wave functions $\psi_m(x)$ and $\psi_n(x)$ are said to be orthogonal if

(A)
$$\int \psi_m(x)\psi_n(x)dx = 0$$

(B)
$$\int \psi_m^*(x)\psi_n(x)dx = 0$$

(C)
$$\int \psi_m(x)\psi_n(x)dx = 1$$

(D)
$$\int \psi_m^*(x)\psi_n(x)dx = 1$$

- 9. Dirac's relativistic theory predicts the existence of
 - (A) proton
 - (B) neutrino
 - (C) positron
 - (D) electron

- 10. The change in frequency due to Doppler Effect does not depend on
 - (A) the speed of the source
 - (B) the speed of the observer
 - (C) the frequency of the source
 - (D) the separation between the source and the observer
- 11. For a perfect black body, the absorptive power is
 - (A) 0
 - (B) ∞
 - (C) 1
 - (D) 0.5
- 12. In Rayleigh scattering of light, the intensity of the scattered light is proportional to
 - (A) $\frac{1}{\lambda^4}$ (B) $\frac{1}{\lambda^2}$ (C) λ^2
 - (D) λ^3
- 13. Two monochromatic light beams of intensities *I* and 4*I* are superposed. The maximum and minimum possible intensities in the resulting beam are
 - (A) 5I and I
 - (B) 5I and 3I
 - (C) 9I and I
 - (D) 9*I* and 3*I*

14. A superconductor is a material

- (A) with zero resistance
- (B) with zero resistance and exhibits perfect diamagnetism
- (C) which shows perfect diamagnetism
- (D) which is ceramic in nature

- 15. A diamagnetic substance has
 - (A) spontaneous magnetization in the absence of applied magnetic field
 - (B) a negative magnetic susceptibility
 - (C) a temperature dependent magnetic susceptibility
 - (D) a net magnetization when applied field H = 0
- 16. In an intrinsic semiconductor, if the energy gap is 0.7 eV, the Fermi level will have an energy of
 - (A) 1.4 eV
 - (B) 1.1 eV
 - (C) 0.35 eV
 - (D) 0.1 eV

17. If I_o is the peak value of the alternating current, then its root mean square (*rms*) value is given by

- (A) $0.637 I_o$
- (B) $0.707 I_o$
- (C) $\frac{I_o}{\pi}$
- (D) $\sqrt{I_o^2}$
- 18. The circuit element used to produce a steady *dc* voltage from the rectified *ac* voltage is called a
 - (A) transformer
 - (B) filter
 - (C) modulator
 - (D) capacitor

19. The efficiency of the Carnot's engine working between the steam point and ice point would be

- (A) Zero
- (B) 100%
- (C) 50%
- (D) 26.8%

- 20. If the latent heat of ice is 80 cal/gm, the increase in entropy when 1 gm of ice at 0° C is converted into water at 0° C
 - (A) 0.29 cal/gm
 - (B) Zero
 - (C) 273 cal/gm
 - (D) 80 cal/gm
- 21. One Curie (unit of radioactivity) is equal to
 - (A) 3.7×10^{10} disintegrations per second
 - (B) 3×10^7 disintegrations per second
 - (C) 3×10^{10} disintegrations per second
 - (D) 3.7×10^7 disintegrations per second
- 22. If *r* is the radius of the nucleus and *A* is the atomic weight (mass number), then
 - (A) *r* is proportional to $A^{1/3}$
 - (B) *r* is proportional to $A^{1/2}$
 - (C) r is proportional to $\frac{A}{2}$
 - (D) *r* is inversely proportional to $A^{1/3}$
- 23. If the radius of the nucleus is of the order of 10^{-13} cm and the mass of the nucleus is of the order of 10^{-24} gm, then the density of the nuclear matter is of the order of
 - (A) 10^{8} gm/cc (B) 10^{20} gm/cc (C) 10^{14} gm/cc (D) 10^{12} gm/cc
- 24. The mass of the neutron is
 - (A) same as the mass of the proton
 - (B) slightly lesser than the mass of the proton
 - (C) slightly greater than the mass of the proton
 - (D) less than the mass of the electron

- 25. The penetrating powers of α , β and γ radiation in decreasing order are
 - (A) γ , α , β
 - (B) γ , β , α
 - $(C) \quad \alpha \,,\, \beta,\, \gamma$
 - (D) β, γ, α
- 26. The essential requirement for a molecule to give rise to rotational Raman spectrum is that
 - (A) The molecule must have a permanent electric dipole moment
 - (B) The molecule must have a center of symmetry
 - (C) The molecule must have an anisotropic polarizability
 - (D) The molecule must be in gaseous form only
- 27. Which one of the following nuclei **CANNOT** be studied by Nuclear Magnetic Resonance (NMR) Spectroscopy?
 - (A) ¹H
 - (B) ${}^{14}N$
 - (C) ⁷Li
 - (D) ^{12}C
- 28. The quadrupole moment of a nucleus has its origin in
 - (A) Electrostatic interaction of the nucleus with its environment
 - (B) Spherically symmetric distribution of nuclear charge in the nuclear volume
 - (C) Interaction of the nucleus with external magnetic field
 - (D) Non spherically symmetric distribution of nuclear charge in the nuclear volume
- 29. The output of the Ruby laser is a
 - (A) pulsed wave
 - (B) continuous wave
 - (C) saw tooth wave
 - (D) triangular wave
- 30. The particles obeying Maxwell-Boltzmann statistics are
 - (A) Identical and indistinguishable
 - (B) Identical but distinguishable
 - (C) Photons
 - (D) Fermions

- 31. In 1.5 years half of 128 mg of radioactive isotope decays. The amount present, after 6 years is
 - (A) 2 mg
 - (B) 4 mg
 - (C) 8 mg
 - (D) 16 mg

32. The number of radial nodes in 3p orbitals are

- (A) 1
- (B) 2
- (C) 3
- (D) 4

33. The nature of the bond between H and Cl in hydrogen chloride is

- (A) covalent bond
- (B) ionic bond
- (C) co-ordinate bond
- (D) covalent in gaseous state and ionic in solution
- 34. The cathode used in polaragraphy
 - (A) DME
 - (B) Hg pool
 - (C) Pt electrode
 - (D) Zinc electrode
- 35. The bond order of nitrogen molecule is
 - (A) 2
 - (B) 2.5
 - (C) 3
 - (D) 3.5
- 36. DMSO is..... solvent.
 - (A) non-polar
 - (B) protic
 - (C) polar protic
 - (D) dipolar aprotic

- 37. Which of the following is a secondary cell?
 - (A) Leclanche cell
 - (B) Lead storage battery
 - (C) Concentration cell
 - (D) Fuel cell
- 38. During the decomposition of H_2O_2 to give dioxygen, 48 g O_2 is formed per minute at a certain point of time. The rate of formation of water at this point is
 - (A) $0.75 \text{ mol min}^{-1}$
 - (B) 1.5 mol min^{-1}
 - (C) $2.25 \text{ mol min}^{-1}$
 - (D) 3.0 mol min^{-1}
- 39. The number of carbon atoms per unit cell of the diamond is
 - (A) 8
 - (B) 6
 - (C) 1
 - (D) 4

40. In calcium fluoride, having the fluorite structure the coordination number of Ca^{2+} ion and F^{-} ion are

- (A) 4 and 2
- (B) 6 and 6
- (C) 8 and 4
- (D) 4 and 8
- 41. The IUPAC name of the compound is



- (A) 2-oxobutanoic acid
- (B) 3-oxobutanoic acid
- (C) 3-oxo-2-propanoic acid
- (D) 2-oxopropan-1-oic acid

- 42. Which among the following aminoacids will give yellow colour in Ninhydrin test?
 - (A) Proline
 - (B) Histidine
 - (C) Cystine
 - (D) Serine
- 43. Baeyer Villiger rearrangement of benzophenone will give
 - (A) phenyl benzoate
 - (B) diphenyl carbinol
 - (C) benzil
 - (D) 2-phenylphenol
- 44. The rearrangement involved in Arndt Eistert synthesis is
 - (A) Lossen rearrangement
 - (B) Curtius rearrangement
 - (C) Wolff rearrangement
 - (D) Wagner Meerwin rearrangement
- 45. Which of the following undergoes SN1 solvolysis at a faster rate in 60% ethanol?
 - (A) Bromomethane
 - (B) Bromoethane
 - (C) 2-Bromopropane
 - (D) 2-Bromo-2-methylpropane
- 46. Which of the following compounds has the highest pKa value in water?
 - (A) HCOOH
 - (B) CH₃CH₂COOH
 - (C) (CH₃)₃CCOOH
 - (D) CICH₂COOH
- 47. Which of the following reagents is commonly used for peptide bond formation?
 - (A) DCC
 - (B) DDQ
 - (C) LDA
 - (D) SeO₂

- 48. The synthetic equivalent for phenyl anion is
 - (A) PhBr
 - (B) PhN₂Cl
 - (C) PhMgBr
 - (D) PhNO₂

49. Which of the following statements is **CORRECT**?

- (A) Nicotine is an alkaloid
- (B) Nicotine is a terpenoid
- (C) Nicotine is a steroid
- (D) Nicotine is a protein

50. How many primary hydroxyl groups does glucose have?

- (A) 5
- (B) 0
- (C) 1
- (D) 4

51. A compound with the molecular formula $C_5H_{12}O$ cannot be

- (A) a primary alcohol
- (B) an ether
- (C) a secondary alcohol
- (D) a ketone
- 52. Which one is inorganic benzene?
 - (A) $B_3N_3H_6$
 - (B) B₂H₆
 - (C) B_2H_4
 - (D) B_2H_{10}
- 53. Among the following which has the lowest electronegativity?
 - (A) C
 - (B) Cs
 - (C) Be
 - (D) Zn

54. The shape of XeF_4 is

- (A) tetrahedral
- (B) square planar
- (C) pyramidal
- (D) linear

55. Most abundant element in earth's crust is

- (A) Si
- (B) Al
- (C) Zn
- (D) Fe

56. The magnetic moment of Mn^{2+} ion is

- (A) 5.92 BM
- (B) 2.80 BM
- (C) 8.95 BM
- (D) 3.90 BM

57. Copper acetate monohydrate dimer is

- (A) ferromagnetic
- (B) ferrimagnetic
- (C) antiferromagnetic
- (D) diamagnetic

58. Frenkel defect is shown by which of the following solids?

- (A) NaCl
- (B) KBr
- (C) AgBr
- (D) FeO

59. The d orbitals split in a square planar complex as

- (A) $dx^2 y^2 > dxy > dz^2 > dxz = dyz$
- (B) $dx^2 y^2 > dxy > dxz = dyz > dz^2$
- (C) $dxy > dx^2 y^2 > dxz = dyz > dz^2$
- (D) $dxy = dx^2 y^2 > dxz = dyz > dz^2$

60. The crystal field stabilization energy of the complex, $[COCl_4]^{2-}$ is

- (A) 12 Dq
- (B) 8 Dq
- (C) 12 Dq + 2p
- (D) 6 Dq + p

61. The intestinal villi are mostly present in abundance in the

- (A) Duodenum
- (B) Ileum
- (C) Jejunum
- (D) Large Intestine

62. What is the tidal volume of respiration?

- (A) 3,000 ml
- (B) 1,100 ml
- (C) 500 ml
- (D) 1,200 ml
- 63. Henle's loop lies in the
 - (A) Kidney
 - (B) Renal Medulla
 - (C) Renal Cortex
 - (D) Renal column of Bertini
- 64. The Reflex centre controlling the musculature of the body is
 - (A) Cerebellum
 - (B) Spinal cord
 - (C) Frontal lobe

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(D) Basal Ganglia

65. Which hormone is responsible for ovulation?

- (A) FSH
- (B) LTH
- (C) LH
- (D) Progesterone

- 66. Which of the following is a muscular tube as far as male reproductive system of man is concerned?
 - (A) Seminiferous tube
 - (B) Epididymis
 - (C) Corpus cavernosum
 - (D) Vas deferens
- 67. Colour blind people cannot distinguish between
 - (A) Red and Blue
 - (B) Red and Grey
 - (C) Red and Yellow
 - (D) Red and Green
- 68. The unique process in Oogenesis is
 - (A) Production of haploid nucleus
 - (B) Production of mature gamete
 - (C) Acquisition of food reserves
 - (D) Formation of spermatid
- 69. The main contributors of acid rain are
 - (A) Sulphur oxides and carbon oxides
 - (B) Nitrogen oxides and sulphur oxides
 - (C) Carbon dioxide and carbon monoxide
 - (D) Carbon and nitrogen oxides
- 70. Filaria is caused by the infections of
 - (A) *Plasmodium vivax*
 - (B) Wuchereria bancrofti
 - (C) Trypanosoma gambiense
 - (D) Entamoeba histolytica

71. Which of the following cell organelle is involved in the breakdown of organic matter?

- (A) Lysosome
- (B) Mitochondria
- (C) Cytoplasm
- (D) Golgi bodies

- 72. Flagellated cells, which line the spongocoel in poriferans are known as
 - (A) Ostia
 - (B) Mesenchymal cells
 - (C) Choanocytes
 - (D) Oscula
- 73. A characteristic feature, which is present only in the phylum Coelenterata, is
 - (A) Nematocytes
 - (B) Flame cells
 - (C) Hermaphroditism
 - (D) Polymorphism
- 74. The swollen bottom part of the carpel of a flower is called as
 - (A) Ovary
 - (B) Style
 - (C) Stigma
 - (D) Anther
- 75. Which of the following plants can survive in extreme saline conditions?
 - (A) Epiphytes
 - (B) Halophytes
 - (C) Xenophytes
 - (D) Hydrophytes
- 76. Which of the following is **NOT** a component of chlorophyll?
 - (A) Calcium
 - (B) Carbon
 - (C) Magnesium
 - (D) Hydrogen
- 77. The food conducting tissue of plants is
 - (A) Phloem
 - (B) Xylem
 - (C) Collenchyma
 - (D) Parenchyma

- 78. Organisms having multiple genomes over their basic number are called
 - (A) Euploids
 - (B) Aneuploids
 - (C) Haploids
 - (D) Polyploids
- 79. The sequential phase of cell cycle is
 - (A) S G1 G2 M
 - (B) G1 S G2 M
 - $(C) \quad G1 \ G2 \ S \ M \ D$
 - (D) M G1 S G2
- 80. The most active phase in cell division is
 - (A) Prophase
 - (B) Anaphase
 - (C) Metaphase
 - (D) Telophase
- 81. Which of the following is cytosolic nucleic acid?
 - (A) m-RNA
 - (B) r-RNA
 - (C) t-RNA
 - (D) DNA
- 82. VAM fungi are used to supply which of the following for plant growth?
 - (A) Phosphorus
 - (B) Potassium
 - (C) Nitrogen
 - (D) Magnesium

83. The term Epicalyx refers to the whorl of

- (A) Sepals
- (B) Petals
- (C) Stamens
- (D) Bracteoles

- 84. The heart of cockroach is
 - (A) 13 chambered
 - (B) Single chambered
 - (C) 10 chambered
 - (D) 7 chambered

85. Number of chromosomes in Down's syndrome is

- (A) 46
- (B) 47
- (C) 48
- (D) 49

86. Study of ageing is known as

- (A) Palynology
- (B) Gerontology
- (C) Genetics
- (D) Speraelogy
- 87. Mesozoic era was dominated by
 - (A) Reptiles
 - (B) Birds
 - (C) Fishes
 - (D) Mammals
- 88. Which one of the following is **NOT** a living fossil?
 - (A) Peripatus
 - (B) King crab
 - (C) Sphenondon
 - (D) Archeopteryx
- 89. The lifespan of RBC is
 - (A) 12 days
 - (B) 120 days
 - (C) 16 days
 - (D) 150 days

- 90. The infective stage of Plasmodium to man is
 - (A) Merozoite
 - (B) Schizont
 - (C) Sporozoite
 - (D) Trophozoite
- 91. Which is followed to maintain and document the chronological history of the evidence?
 - (A) Chain of custody
 - (B) Documentation
 - (C) Videography
 - (D) Photo log
- 92. Which of the following poisons produces Cherry-red colour on dead body?
 - (A) Carbon monoxide
 - (B) Cyanide
 - (C) Pesticide
 - $(D) \quad Both \ (A) \ and \ (B)$
- 93. Name the animal anesthetic widely being abused as a narcotic drug
 - (A) MDMA
 - (B) Ketamine
 - (C) Amphetamine
 - (D) LSD
- 94. According to Locard's principle which of the following is correct?
 - (A) Similar objects can be compared
 - (B) Every object changes with the passage of time
 - (C) Every contact leaves some trace
 - (D) Every object is unique

95. Crystals produced in Teichmann test are

- (A) Brown feathery
- (B) Brown rhombic
- (C) Red feathery
- (D) Red rhombic

- 96. Name a very strong protein which resists the decomposition of hair
 - (A) Melanin
 - (B) Keratin
 - (C) Gelatin
 - (D) Fibroin

97. A study of characteristics of different people and their relationships is

- (A) Ethnology
- (B) Ecology
- (C) Monospecific
- (D) Monoecious
- 98. Radioactivity of a substance is measured by
 - (A) TCD detector
 - (B) MCT detector
 - (C) EC detector
 - (D) Geiger Muller counter
- 99. 'Joule Burn' is associated with
 - (A) Arson
 - (B) Fire occurrence
 - (C) Electrocution
 - (D) Asphyxia
- 100. The technique, which separates the mixture into its components is
 - (A) Chromatography
 - (B) Immunoassay
 - (C) Micro diffusion test
 - (D) Spectroscopy

101. Double helical model of DNA was elucidated by

- (A) Alec Jeffrey
- (B) Edmond Locard
- (C) William Herschel
- (D) Watson and Crick

- 102. Preservation of foot print on snow is done by
 - (A) Paster of Paris cast
 - (B) Sulphur cast
 - (C) Mud cast
 - (D) Wax cast

103. Lower case, joined, running handwriting is known as

- (A) Cursive
- (B) Disconnected
- (C) Graphic
- (D) Script

104. Main component of MERMER (Memory and Encoding Related Multifaceted Electroencephalographic Response) is

- (A) P 150
- (B) P 300
- (C) P 450
- (D) P 600
- 105. Bullets which are designed to expand on impact is called
 - (A) Dum Dum bullet
 - (B) Richochet bullet
 - (C) Souvenier bullet
 - (D) Piggy back bullet
- 106. Luminol test is used to detect
 - (A) Gun powder residues
 - (B) Latent finger prints
 - (C) Seminal stains
 - (D) Blood stains
- 107. Drug used in Narco-analysis is
 - (A) Barium sulphate
 - (B) Magnesium bircarbonate
 - (C) Potassium sulphate
 - (D) Sodium amytal

108. Hacking is defined under

- (A) Section 66 Indian IT Act 2000
- (B) Section 70 Indian IT Act 2000
- (C) Section 75 Indian IT Act 2000
- (D) Section 86 Indian IT Act 2000
- 109. refers to sending email to thousands and thousands of users similar to a chain letter
 - (A) Trojan attack
 - (B) Email spoofing
 - (C) Email bombing
 - (D) Email spamming
- 110. Act of determining whether a particular user or a computer system has the right to carry out certain activity is
 - (A) authentication
 - (B) authorization
 - (C) non-repudiation
 - (D) None of the above
- 111. The branch of forensic which uses insects as evidence to help reconstruct the circumstances
 - (A) Forensic chemistry
 - (B) Forensic entomology
 - (C) Forensic odontology
 - (D) Forensic toxicology
- 112. Widmark's formula is used for measurement of blood levels of
 - (A) alcohol
 - (B) arsenic
 - (C) barbiturates
 - (D) benzodiazepines
- 113. Latent fingerprints can be detected by
 - (A) Benzidine test
 - (B) Brentamine fast blue test
 - (C) Isoenzyme marker
 - (D) Ninhydrin reagent

- 114. Faeces stains are identified by odour, presence of indigested matter, vegetable fibres and
 - (A) stercobilin
 - (B) choline iodide
 - (C) spermine picrate
 - (D) hemochromogen
- 115. Dermal nitrate test is used to detect
 - (A) gun powder residues
 - (B) latent finger prints
 - (C) seminal stains
 - (D) old blood stains
- 116. The science of writing or solving codes is
 - (A) Calligraphy
 - (B) Cryptography
 - (C) Steganography
 - (D) Seismography
- 117. Anthropology as a system of identification was developed by
 - (A) Alphonse Bertillon
 - (B) Orfila
 - (C) William West
 - (D) Maples
- 118. Choose from the following, the commonest preservative used for viscera in all cases of poisoning including carbolic acid poisoning
 - (A) Ethylene glycol
 - (B) Formalin
 - (C) Cresol
 - (D) Saturated solution of sodium chloride
- 119. Klinefelter's syndrome is characterized by following chromosomal pattern
 - (A) XXY
 - (B) XO
 - (C) XXX
 - (D) XYY

120. If both mother and father are AB blood group then child can never be

- (A) A
- (B) B
- (C) AB
- (D) O

121. Decision Support Systems (DSS) is

- (A) A family of relational database management systems marketed by IBM
- (B) Interactive systems that enable decision makers to use databases and models on a computer in order to solve ill-structured problems
- (C) It consists of nodes and branches starting from a single root node. Each node represents a test, or decision
- (D) None of the above

122. Which of the following are procedural languages?

- (A) Pascal
- (B) Smalltalk
- (C) C
- $(D) \quad Both \ (A) \ and \ (C)$
- 123. Which of the following is a type of non-impact printer?
 - (A) Daisy Wheel Printer
 - (B) Dot Matrix Printer
 - (C) Both Dot Matrix Printer and Daisy Wheel Printer
 - (D) Laser printer
- 124. Roman number system is a
 - (A) Positional number system
 - (B) Non-positional number system
 - (C) Both (A) and (B)
 - (D) None of the above

125. The time required to move the disk head to the desired track is known as

- (A) Access Time
- (B) Latency Time
- (C) Seek Time
- (D) Track Time

- 126. Which of the following is an attribute that can uniquely identify a row in a table?
 - (A) Secondary key
 - (B) Candidate key
 - (C) Foreign key
 - (D) None of the above
- 127. Factors which affect the performance of learner system does not include
 - (A) training scenario
 - (B) type of feedback
 - (C) good data structures
 - (D) learning algorithm
- 128. Which one of the following statements is FALSE regarding "Cookies"?
 - (A) Cookies are programs which run in the background of web-client
 - (B) Cookies have the potential of being used to violate the privacy of users
 - (C) Cookies cannot contain more than 4Kb of data
 - (D) Cookies usually contain data in the form of name = value pairs
- 129. When trying to access a URL, the following message is displayed on the browser:

Server; Error 403

What could be the reason for the massage?

- (A) The requested HTML file is not available
- (B) The path to the interpreter of the script file is invalid
- (C) The first line of the output form the script is not a valid HTTP header
- (D) The requested HTML file or CGI script has insufficient permission
- 130. Which of the following is the collection of data objects that are similar to one another within the same group?
 - (A) Partitioning
 - (B) Grid
 - (C) Cluster
 - (D) Table
- 131. supports characters for all known languages.
 - (A) ASCII
 - (B) EBCDIC
 - (C) UNICODE
 - (D) GRAYCODE

- 132. The registers found in the processor unit are
 - (A) Operational Registers
 - (B) Memory Registers
 - (C) Storage Registers
 - (D) Binary Registers
- 133. System Software usually includes a program called a, which helps the programmer to find errors in a program.
 - (A) Write Buffer
 - (B) Read Buffer
 - (C) Debugger
 - (D) Both (A) and (C)
- 134. Cache memory enhances
 - (A) Memory Capacity
 - (B) Memory Access Time
 - (C) Secondary Storage Access Time
 - (D) Data Transfer Time
- 135. What is the purpose of RAID system?
 - (A) It increases the processor speed
 - (B) It increases operating system efficiency
 - (C) It increases the disk storage capacity and availability
 - (D) It decreases operating system efficiency
- 136. What is the units for measuring the CPU performance?
 - (A) BPS
 - (B) MIPS
 - (C) MHz
 - (D) MFLOP
- 137. 'OS' computer abbreviation usually means?
 - (A) Order of Significance
 - (B) Open Software
 - (C) Operating System
 - (D) Optical Sensor

- 138. '.MOV' extension refers usually to what kind of file?
 - (A) Image file
 - (B) Animation/movie file
 - (C) Audio file
 - (D) MS Office document

139. What is a URL?

- (A) A computer software program
- (B) A type of UFO
- (C) The address of a document or "page" on the World Wide Web Uniform Resource Locator
- (D) An acronym for Uniform Resources Learning
- 140. What do we call a collection of two or more computers that are located within a limited distance of each other and that are connected to each other directly or indirectly?
 - (A) Internet
 - (B) Intranet
 - (C) Local Area Network
 - (D) Wide Area Network
- 141. Which of the following is **NOT** a computer language?
 - (A) Windows 10
 - (B) JAVA
 - (C) PYTHON
 - (D) C++
- 142. Which one of the following manages both current and historic transactions?
 - (A) Online Analytical Processing (OLAP)
 - (B) Online Transaction Processing (OLTP)
 - (C) Spread Sheet
 - (D) Extensible Markup Language (XML)
- 143. Which of the following is a valid e-mail address?
 - (A) support.at.objectivebooks
 - (B) www.objectivebooks.com
 - (C) support@objectivebooks.com
 - (D) support@objectivebooks

144. Expansion of SMTP is

- (A) Single Mail Transfer Protocol
- (B) Simple Message Transfer Protocol
- (C) Simple Message Transmission Protocol
- (D) Simple Mail Transfer Protocol

145. The speed of the net access is defined in terms of

- (A) RAM
- (B) MHz
- (C) Kbps
- (D) Megabytes
- 146. Identify the 2's complement of 101010101.
 - (A) 010101010
 - (B) 010101000
 - (C) 010101011
 - (D) 110101011

147. What's a web browser?

- (A) A kind of spider
- (B) A computer that stores WWW files
- (C) A person who likes to look at websites
- (D) A software program that allows you to access sites on the World Wide Web
- 148. RAM stands for
 - (A) Readly Access Memory
 - (B) Random Access Memory
 - (C) Random Accent Memory
 - (D) Read Access Memory

149. Which of the following is a search engine?

- (A) FTP
- (B) Google
- (C) Netscape
- (D) ARPANET

150. How many bits is a byte?

- (A) 4
 (B) 8
 (C) 16
 (D) 32

FINAL ANSWER KEY									
Subject Name: 625 MSc FORENSIC SCIENCE									
SI No.	Key	SI No.	Key	SI No.	Key	SI No.	Key	SI No.	Key
1	D	31	С	61	В	91	A	121	В
2	В	32	А	62	С	92	D	122	D
3	D	33	А	63	В	93	В	123	D
4	Α	34	А	64	С	94	C	124	В
5	Α	35	С	65	С	95	В	125	С
6	В	36	D	66	D	96	В	126	В
7	Α	37	В	67	D	97	А	127	С
8	В	38	D	68	С	98	D	128	А
9	D	39	Α	69	В	99	С	129	D
10	D	40	С	70	В	100	А	130	С
11	С	41	A	71	А	101	D	131	С
12	Α	42	A	72	С	102	В	132	Α
13	С	43	A	73	А	103	А	133	С
14	В	44	C	74	А	104	В	134	В
15	В	45	D	75	В	105	А	135	С
16	C	46	C	76	Α	106	D	136	С
17	В	47	Α	77	А	107	D	137	С
18	В	48	С	78	D	108	А	138	В
19	D	49	Α	79	В	109	D	139	С
20	A	50	С	80	А	110	В	140	C
21	A	51	D	81	С	111	В	141	Α
22	A	52	Α	82	А	112	Α	142	Α
23	Α	53	В	83	D	113	D	143	C
24	C	54	В	84	Α	114	Α	144	D
25	В	55	Α	85	В	115	А	145	С
26	C	56	A	86	В	116	В	146	С
27	D	57	C	87	А	117	А	147	D
28	D	58	C	88	D	118	D	148	В
29	Α	59	A	89	В	119	А	149	В
30	В	60	Α	90	С	120	D	150	В